
Chapter 1: Introduction

(Sections 1.5 and 1.6)

Lecture 1

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1-5] Units, Standards, and the SI System

When we measure the height of a table we may write 2 m.

result $\xrightarrow{\text{unit}}$ unit which is the meter.
of measurement

Usually, physical quantities have units like:

Quantity	Unit
mass	kg or gram, ..
velocity	m/s or cms, ..
Temperature	Celsius, Fahrenheit, ..

But what unit should we use?

Système International
in French

In the International System (SI system)

The units for Length, Mass, and Time are:

Length : meter (m)

Mass : kilogram (kg)

Time : second (s)

} before this was called
the mks system
meter / / second
kilogram

We shall mostly use the SI system (mks).

Sometime people use the cgs system
centimeter for length / gram for mass / second for time

It is easy to change from mks \rightleftarrows cgs. We can use:

$$1 \text{ m} = 100 \text{ cm}, \quad 1 \text{ kg} = 1000 \text{ gram}$$

Why are length(L), mass(M), time(T) are called base units?

consider the following

$$\begin{array}{lll} \text{Velocity} & \text{m/s} & \Rightarrow L/T \\ \text{Acceleration} & \text{m/s}^2 & \Rightarrow L/T^2 \\ \text{Density} & \text{kg/m}^3 & \Rightarrow M/L^3 \end{array} \left. \begin{array}{l} \text{Note that velocity,} \\ \text{acceleration and density} \\ \text{are expressed in terms of} \\ L, M, T. \end{array} \right\}$$

Similarly, momentum ($p = mv$), force ($F = ma$)
are also expressed in terms of L, M, T .

Example: Show that force which has units of
(newton) can be expressed in terms of
the base quantities L, M, T .

$$F = \overset{\text{mass}}{m} \overset{\text{acceleration}}{a}$$

The unit of force is the newton (N).

$$1 \text{ N} = 1 \text{ kg} \cdot 1 \frac{\text{m}}{\text{s}^2} \equiv M \cdot \frac{L}{T^2}.$$

Therefore the newton is a derived quantity, since we can express it in terms of a combination of the base units M, L, T .

In the SI system there are SEVEN base quantities

Quantity	Unit	Unit abbreviation
Length	meter	m
Time	second	s
Mass	kilogram	kg
Electric Current	ampere	A
Temperature	Kelvin	K
Amount of Substance	mole	mol
Luminous Intensity	candela	cd

1-6] Converting Units

The length of a pen is 3 $\overbrace{\text{inches}}^{\text{in.}}$. Express the length of the pen in terms of cm.

$$1 \text{ in.} = 2.54 \text{ cm}$$

$$\Rightarrow 1 = 2.54 \frac{\text{cm}}{\text{in.}}$$

number

∴ length of pen = 3 in. = 3 in. \times $\underbrace{(2.54 \frac{\text{cm}}{\text{in.}})}_{\text{conversion factor}}$

$$= 3 \times 2.54 \text{ cm}$$

$$= 7.62 \text{ cm.}$$

Example: The speed of a car is 100 km/h. Express the speed in terms of m/s.

Note: $1 \text{ km} = 1000 \text{ m} \Rightarrow 1 = 1000 \frac{\text{m}}{\text{km}}$

$$1 \text{ h} = 3600 \text{ s} \Rightarrow 1 = \frac{1}{3600} \frac{\text{h}}{\text{s}}$$

$$\therefore 100 \frac{\text{km}}{\text{h}} = 100 \frac{\cancel{\text{km}}}{\cancel{\text{h}}} \times \left(1000 \frac{\text{m}}{\text{km}} \right) \times \left(\frac{1}{3600} \frac{\cancel{\text{h}}}{\text{s}} \right)$$

$$= 100 \times \frac{10}{36} \frac{\text{m}}{\text{s}} = 100 \times \frac{5}{18} \frac{\text{m}}{\text{s}}$$

$$\approx 27.8 \text{ m/s.}$$

Question: A car is moving at 20 m/s. Express the speed in units of km/h.

Answer: 72 km/h.

Question: The area of a plate is 32 cm². Express the area in units of m².

$$1 \text{ m} = 100 \text{ cm}$$

$$\therefore 1 \text{ m}^2 = 10^4 \text{ cm}^2 \Rightarrow 1 = \frac{1}{10^4} \frac{\text{m}^2}{\text{cm}^2}$$

$$\begin{aligned}
 \therefore 32 \text{ cm}^2 &= 32 \text{ cm}^2 \times \left(\frac{1}{10^4} \frac{\text{m}^2}{\text{cm}^2} \right) \\
 &= \frac{32}{10^4} \text{ m}^2 = 32 \times 10^{-4} \text{ m}^2 \\
 &= 0.0032 \text{ m}^2 .
 \end{aligned}$$

Some Metric (SI) Prefixes

Prefix	Abbreviation	value
tera	T	10^{12}
giga	G	10^9
mega	M	10^6
kilo	K	10^3
milli	m	10^{-3}
micro	μ	10^{-6}
nano	n	10^{-9}
pico	p	10^{-12}

For examples:

$$1000 \text{ m} = 1 \text{ km}.$$

$$5 \times 10^6 \text{ m} = 5 \mu\text{m}$$

$$20 \times 10^6 \text{ kg} = 20 \text{ Mkg} \quad \text{and so on.}$$

You shoud do the following problems from
Sections 1-5 and 1-6 page 18 of the 7th edition
of the text book:

Q13, Q14, Q15