Carbohydrates:

Monosaccharides:

	Structure	Functional group	Cyclic	Properties
Ribose	СНО НСОН НСОН НСОН СН ₂ ОН D-Ribose (Rib)	5 carbons Aldose	CH2OH H H OH OH OH OH OH	-
Glucose	CHO HCOH HOCH HCOH HCOH HCOH CH ₂ OH D-Glucose (Glc)	6 carbons Aldose Right, left, right.	CH ₂ -OH H H H H OH OH H OH α-D-glucopyranose	Mild sweet Blood sugar Essential energy source Found in every disaccharide and polysaccharide
Galactose	CHO HCOH HOCH HOCH HOCH HCOH CH ₂ OH D-Galactose (Gal)	6 carbons Aldose Right,left,left.	β -D-galactopyranose	Hardly tastes sweet Rarely found as single sugar

Mannose	CHO HOCH HOCH HOCH HCOH CH ₂ OH D-Mannose (Man)	Aldose 6 carbons Left left right	α -D-mannopyranose	-
Fructose	$ \begin{array}{c} CH_{2}OH\\ C=0\\ HO-H\\ H-OH\\ H-OH\\ CH_{2}OH\\ D-Fructose \end{array} $	Ketose 6 carbons	HO-CH ₂ 6 1 CH ₂ OH 2 0H 2 OH 0H fructose	Sweetest Fruit and honey Added to soft drinks and desserts.

Modified sugars :

Oxidation (sugar acids) Oxidation of glucose c1 aldhyde —> gluconic acid (gluconate) Weak oxidizing agent Oxidation of c6 OH —> glucuronic acid (glucuronate) Enzymes Oxidation of c1 and c6 —> glucaric acid Strong oxidizing agent	 Reduction 1- Sugar alcohols: sorbitol , mannitol , xylitol 2- Deoxysugars: one or more hydroxyl groups are replaced with hydrogen 	
Esterification (sugar esters)	Glycosides	
Phosphate esters		
When carbon no 1 (anomeric carbon) is esterified it	O glycoside	
turns into phosphoacetal	C glycoside	
When carbon no 6 is esterified, it is turns into		
phosphate ester.	Anomeric carbon !!!	
Amino sugars		
Adding N to any carbon except anomeric Further modification can occur through acetylation	$\begin{array}{c} & HOCH_2 \\ HOCH_2 \\ & 5 \\ & 0 \\ & 4 \\ & 0H \\$	
(Chitin,		

Disaccharides



Oligosaccharides :

1-Raffinose : glucose + galactose + fructose

Found in beans and vegetables Humans lack the enzyme alpha galactose dash that is needed to break raffinose but intestinal bacteria ferment it hydrogen methane and other gases

2- Streptomycin erythromycin (antibiotics)

- **3-Doxorubicin** (chemotherapy)
- 4-Digoxin (cardiovascular disease)

Polysaccharide:

	Monomer	Bonds	Purpose	Homo,hetero	More info	
Glycogen	Glucose	Alpha 1,4 1,6 branches	Storage mammals	Ното	Extensively branched	Alpha bond is flexible
Starch	Glucose	Alpha 1,4 1,6 branches	Storage plants	Homo	Amylose Amylopectin	Alpha (can be bent)
Dextran	Glucose	Alpha 1,6 Branches 1,2 1,3 1,4	Storage yeasts and bacteria	Homo	Branching points are variable	
Cellulose	Glucose	Beta 1,4	Structural plants	Homo	No branches	Beta (not flexible)
Chitin	N-acetyl-b-d- glucosamine	Beta 1,4	Structural Exoskeleton	Homo	Amino sugars	
Pectin	Methyl galactoronate And Galactoronate	Alpha 1,4	Bacteria Plants Jello	Hetero	Sugar acid	
GAGs	Derivatives of amino sugars	All beta except heparin alpha	ECM	Hetero	Contain negatively charged carboxylate or sulfate	Repeated disaccharide
Bacterial cell wall	NAM NAG			Hetero	Rigid	Contains oligopeptide

GAG	Localization	Comments
Hyaluronate	synovial fluid, vitreous humor, ECM of loose connective tissue	the lubricant fluid , shock absorbing As many as 25,000 disaccharide units
Chondroitin sulfate	cartilage, bone, heart valves	most abundant GAG
Heparan sulfate	basement membranes, components of cell surfaces	contains higher acetylated glucosamine than heparin
Heparin	component of intracellular granules of mast cells lining the arteries of the lungs, liver and skin	A natural anticoagulant
Dermatan sulfate	skin, blood vessels, heart valves	
Keratan sulfate	cornea, bone, cartilage aggregated with chondroitin sulfates	Only one not having uronic acid