

فريق طوفان الأقصى

METABOLISM

Modified N.3



nanoschematic

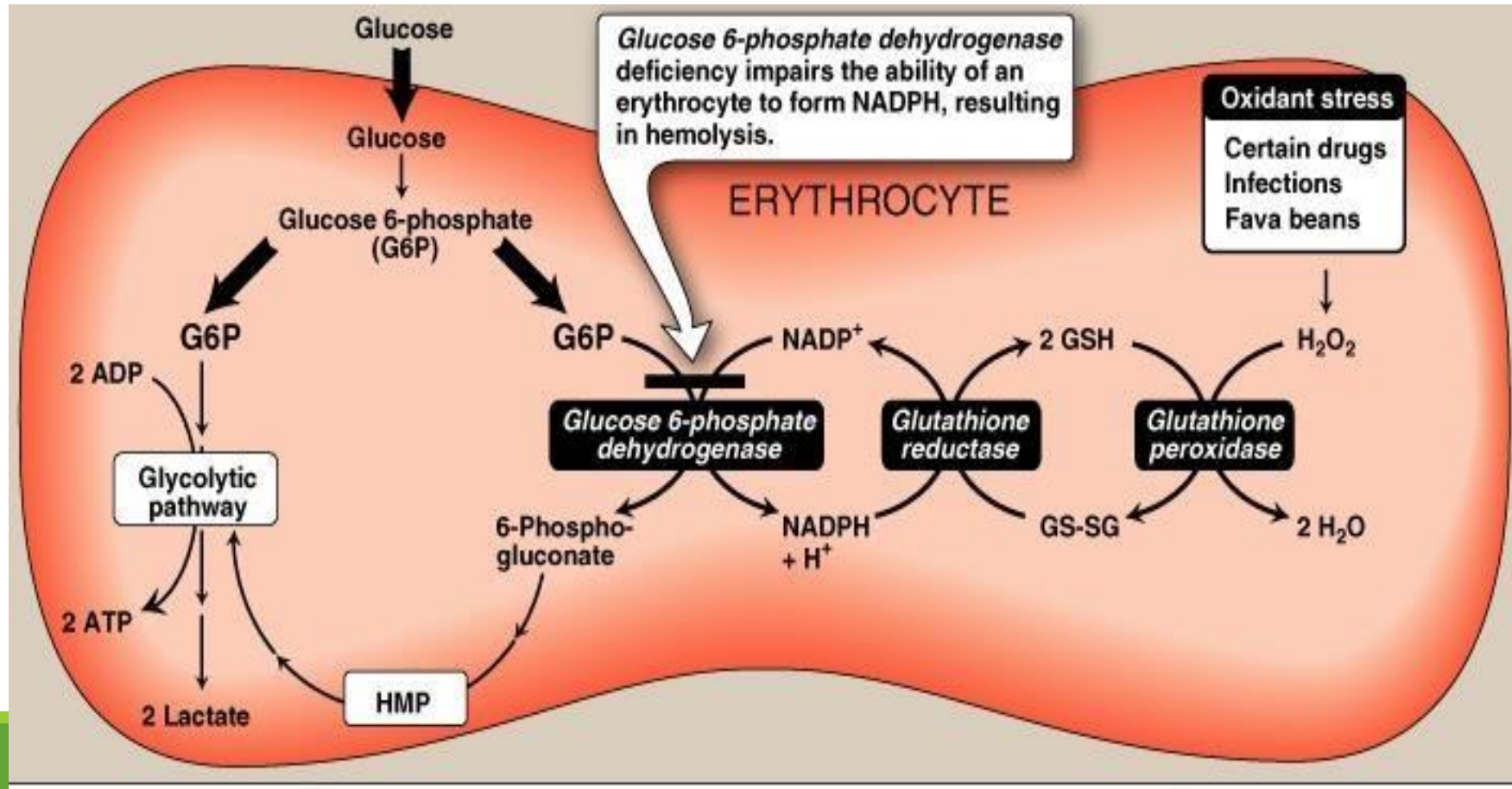
Writer: علاء خضر
صهيب زعيتر

Corrector: علاء خضر
صهيب زعيتر

Role of G6PD in red blood cells



GSH helps maintain the SH groups in proteins in the reduced state
Oxidation → denaturation of proteins and rigidity of the cells



- **The complement in this slide:** we talked about G6PD deficiency and this disease results in reduced production of NADPH.
- More than 400 mutations can cause G6PD deficiency, and every mutation controls the severity of the condition. Therefore, this disease is variable among different patients.
- Those patients will have hemolysis.
- RBC as you remember is depending on glucose as source of energy.
- When glucose concentration is high it's going to be up taken by **ERYTHROCYTE** and then phosphorylated to **glucose 6-phosphate** and continue **glycolytic pathway**, it may also enter **anaerobic respiration** producing **lactate** to provide the low energy demands of this cell.
- In normal conditions this **glucose 6-phosphate** enters **PPP** (pentose phosphate pathway) and the reason of PPP in RBC is the **NADPH**.
- The main purpose is to produce **NADPH** to fight oxidative stress using **Glutathione**.

- In those patients, this pathway is blocked or reduced at an initial stage, which is why a decreased amount of NADPH is produced, leading to the accumulation of more oxidative stress.
- These ROS start to react haphazardly with the surrounding proteins (it might be Na⁺/K⁺ pump **تأثر** عليهم بطريقة تخرب الوظيفة تبعثهم and this will result in more rigidity of the RBC, this will lead to hemolysis (premature death before 120 days))

G6PD Deficiency Variants

■ **NOTE:** this disease has many levels of severity:

- Wild type B
- Mediterranean Variant B⁻ (Class II) : 563C → T
- African Variant A⁻ (Class III); two point mutation
- African Variant A; Normal activity 80%
- Very severe deficiency (Class I)
- Majority missense mutation point mutation
- Large deletions or frame shift; Not Observed

■ The complement in this slide:

• All these 400 mutation most of them are point mutation(nucleotide بدلنا مكانه واحد ثاني)

1- Class I : the most severe mutation which causes the greatest deficiency of the enzyme.

2- Class II (B- variant): cytosine is replaced with thymine on position 563 on the gene and this type is the most common type in Mediterranean countries , this contains severe deficiency of G6PD(the wild type of it called B).

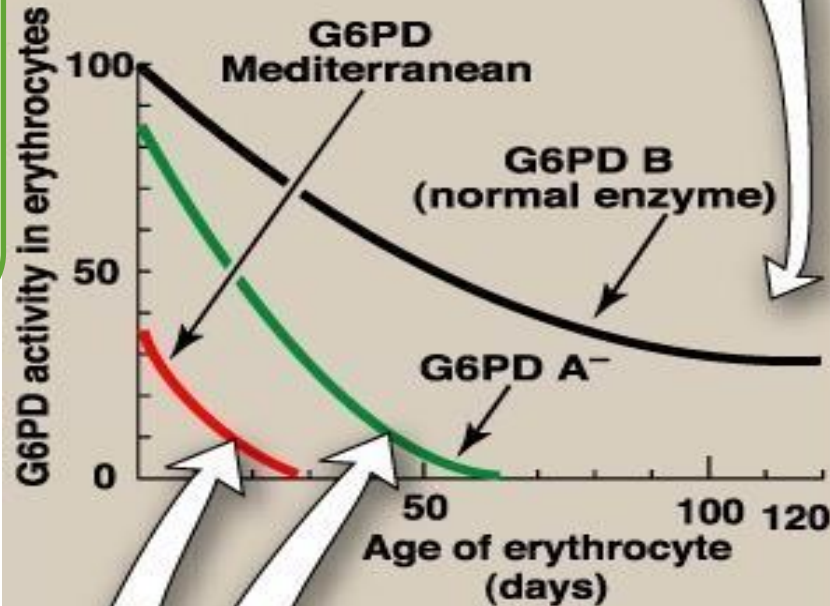
3- Class III (A- Variant): more predominant in African population , their enzyme activity is close to normal, made up from 2-point mutation, however the amount of deficiency is less(when it is affected, we call it A- but when it is normal, we call it A).

4- Class IV: is the least severe, mostly normal, expression of the enzyme is high.

Look at the picture in the next slide.

Classification of G6PD Deficiency Variants

Although the activity of the normal enzyme declines as red cells age, even the oldest cells have a sufficient level of activity to provide protection against oxidative damage and hemolysis.



By contrast, very few *G6PD Mediterranean* red cells have sufficient enzyme activity to prevent oxidative damage, whereas a substantial fraction of young *G6PD A⁻* red cells are able to provide protection.

NOTE: focus on the day of death and the activity

Class	Clinical symptoms	Residual enzyme activity
I	Very severe	<2%
II	Severe	<10%
III	Moderate	10-50%
IV	None	> 60%

NOTE: i need more than %60 of the enzyme for full function of pathway

Enzymes that catalyze antioxidant reactions

■ **NOTE:** Oxidative stress can be handled by NADPH by different molecules like **Glutathione**, another ways:

2. Super oxide dismutase (**SOD**)



- **NOTE:** Super Oxide Dismutase (SOD), acts on super oxide ion and converts it into Hydrogen peroxide
- 2) Hydrogen peroxide (H_2O_2) can be degraded by catalase enzyme into water and oxygen

3. Catalase (in the peroxisome)



■ **NOTE:** Mechanisms from outside the body like diet (Vitamins):
Carotenoids (Vitamin A): Beta carotene get cleavage and it releases Vitamin A.

Anti oxidant chemicals

- Vitamin E, Vitamin C, Carotenoids

- **NOTE:** we said that ROS always will be side product of your metabolism, is that mean that every reaction and every enzyme can produce ROS? **No**
- In specific enzymes whose functions involve oxygen, they are the ones that will produce reactive oxygen species (ROS) as a side product of that reaction.

Sources of ROS in the cell

- Oxidases



Most oxidases produce H_2O_2 (peroxidase)

Oxidases are confined to sites equipped with protective enzymes

- Oxygenases

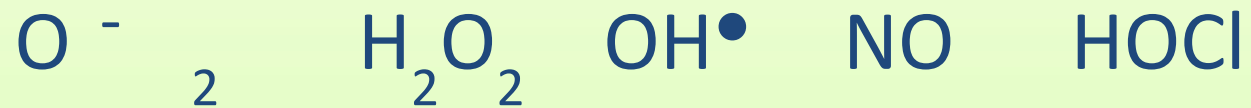
- Mono oxygenases (hydroxylases)

- Dioxygenases in the synthesis of prostaglandins, thromboxanes, leukotrienes

- **The complement in this slide:** the enzymes that are consider as Oxidases (Peroxidase as a sub group) peroxidase phosphate can produce H₂O₂ (substrate react with O₂)
- Oxidases also classify as **Oxidoreductases**.
- Oxygenases also classify as **Oxidoreductases**
- Also by using O₂ and they can be subdivided into:
Monooxygenase (like hydroxylases)
Dioxygenase like: COX 1 AND 2

Sources of ROS in the cell

- Coenzyme Q in Respiratory chain
- Respiratory Burst (during phagocytosis)



- Ionizing Radiation



■ **NOTE:** like UV light, X ray,
nuclear Radiation

- **The complement in this slide: another place that has Coenzyme Q(ubiquinone) in the respiratory chains.**
- **Carries Electrons from Complex 1(NADH dehydrogenase(NADH)) and 2(Succinate dehydrogenase(FADH₂)).**
- **The last electron acceptor in ETC is O₂, so it is an Oxygen rich environment.**
- **When electrons are moving, I might lose some of these electrons and get accepted by the Oxygen to produce Superoxide ion O₂⁻.**
- **Respiratory burst occur during phagocytosis of microorganisms (viruses and bacteria) and their phagocytosis into phagosome that fuses with lysosome (Lysosomes have acidic environment and have same enzymes that can produce a huge amount of ROS/RNOS to lyse and damage the invading microorganisms).**

Cytochrome P450 Mono oxygenase

- Mixed function oxygenase
- Super family of structurally related enzymes



Mitochondrial system

Synthesis by hydroxylation of steroids, bile acids,
active form of Vit. D

Microsomal system

Detoxification of foreign compounds

Activation or inactivation of Drugs

Solubilization to facilitate excretion in urine or feces

■ The complement in this slide:

- It takes an O₂ molecule and use one atom to form an OH group , and another one will interact with H + to form water, the source of H⁺ will be NADPH oxidation.it produces ROS **increasing the oxidative stress.**



- P450 system present in multiple places within the cell , like the **mitochondria** (that is mainly concerned with hydroxylation Reactions which happens during steroids, bile acid and vitamin D synthesis.

Mitochondrial system

Synthesis by hydroxylation of steroids, bile acids,
active form of Vit. D

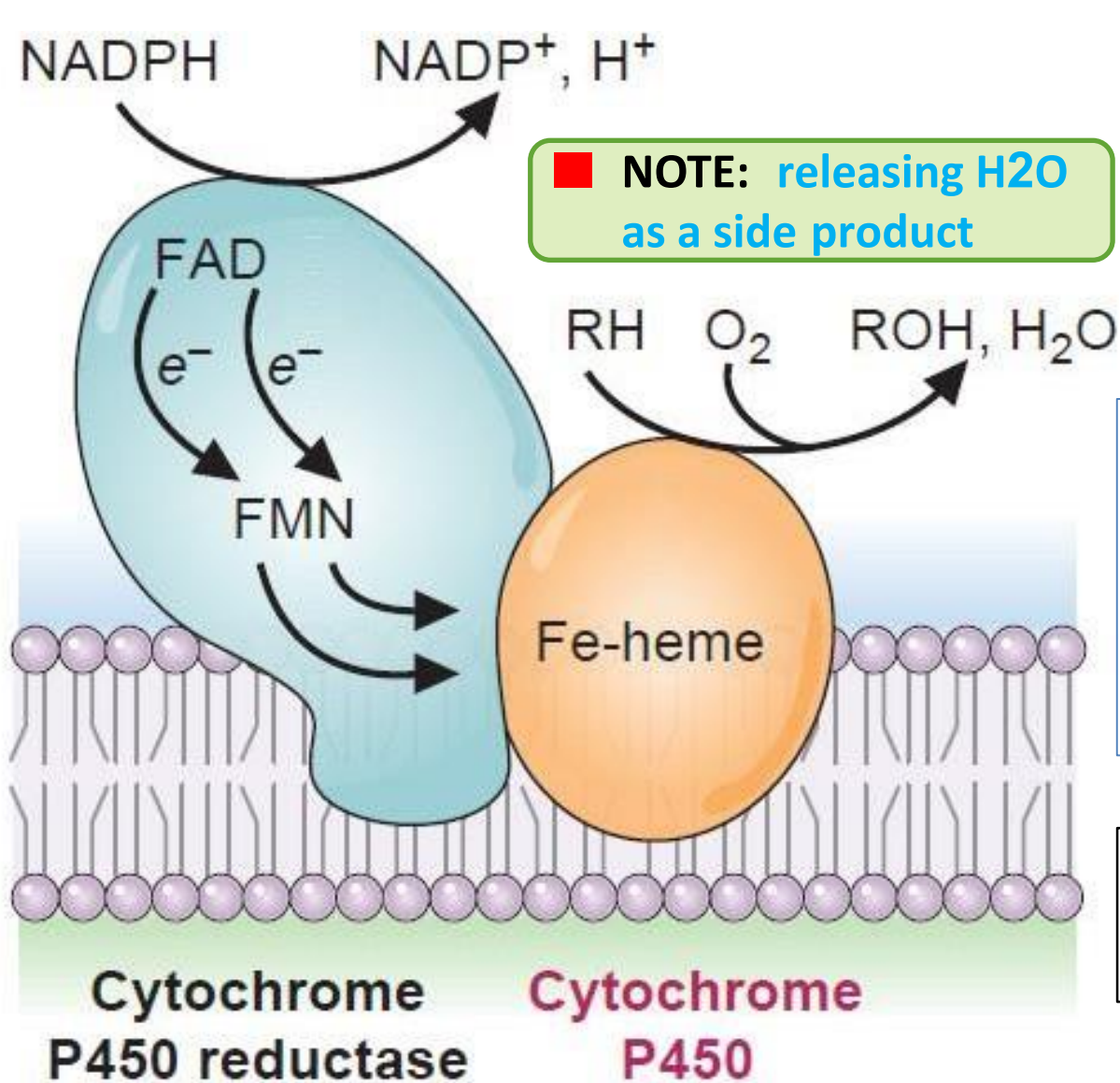
- **Microsomal system** which is present in the ER , its main function is **detoxification of drugs and foreign compounds**, sometimes this system activates the drug. The main purpose of this degradation is to produce **excretable** (mainly in Urine) form of the **drug** (convert it from lipid soluble to water soluble) by **solubilization.**

Microsomal system

Detoxification of foreign compounds

Activation or inactivation of Drugs

Solubilization to facilitate excretion in urine or feces



■ NOTE: releasing H₂O as a side product

Cytochrome P450 Mono oxygenase

Accidental release of free radical intermediates may occur

■ NOTE: Reaction in the picture is the main mechanism.

■ NOTE: the accident when O₂ interacts with electrons --> free radicals

XH₂: electron donor
S: substrate

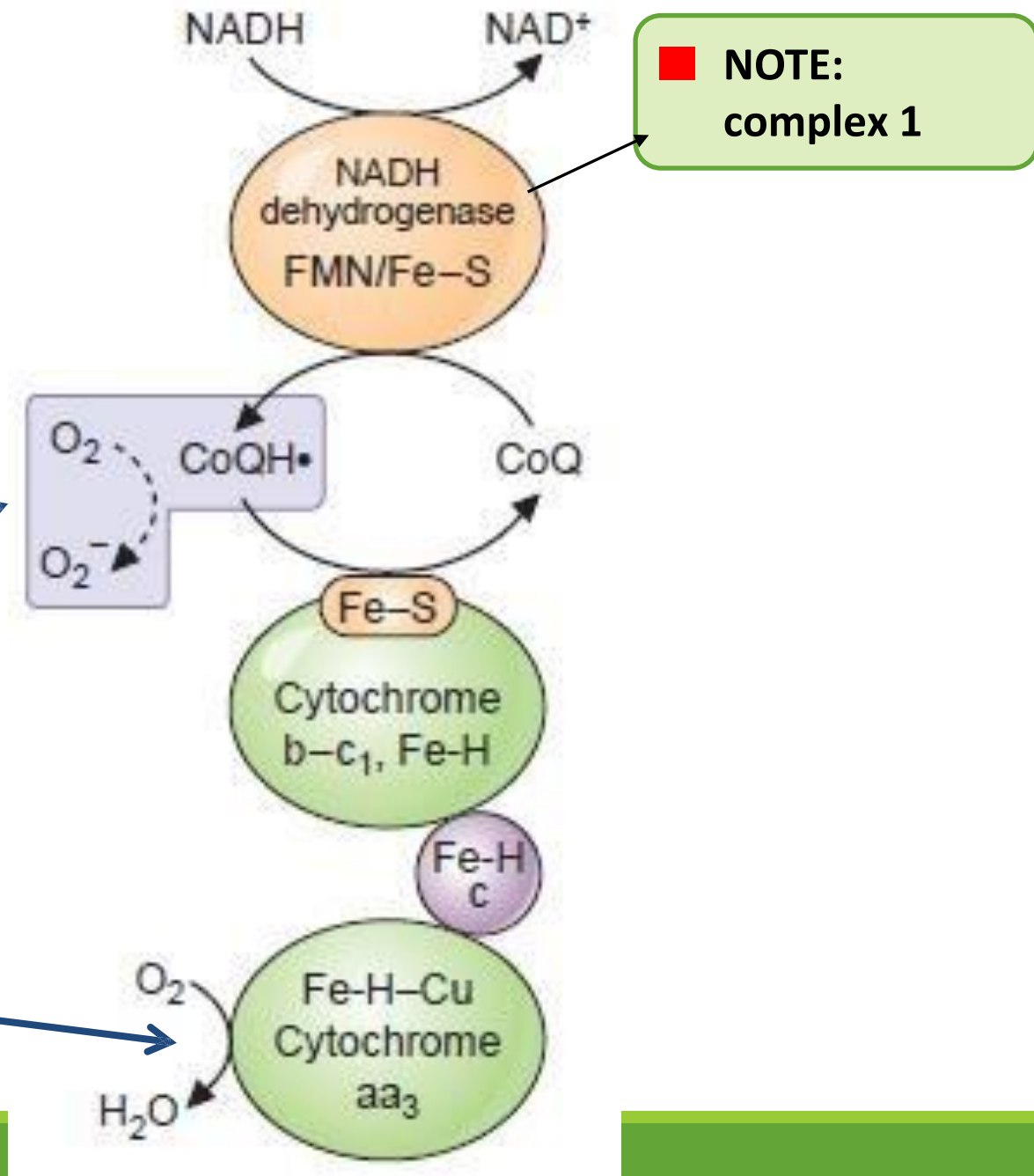


Generation of O_2^- by the respiratory chain

Accidental non-specific interaction

Major source of free radicals

Binuclear center prevents release of free O_2 radicals



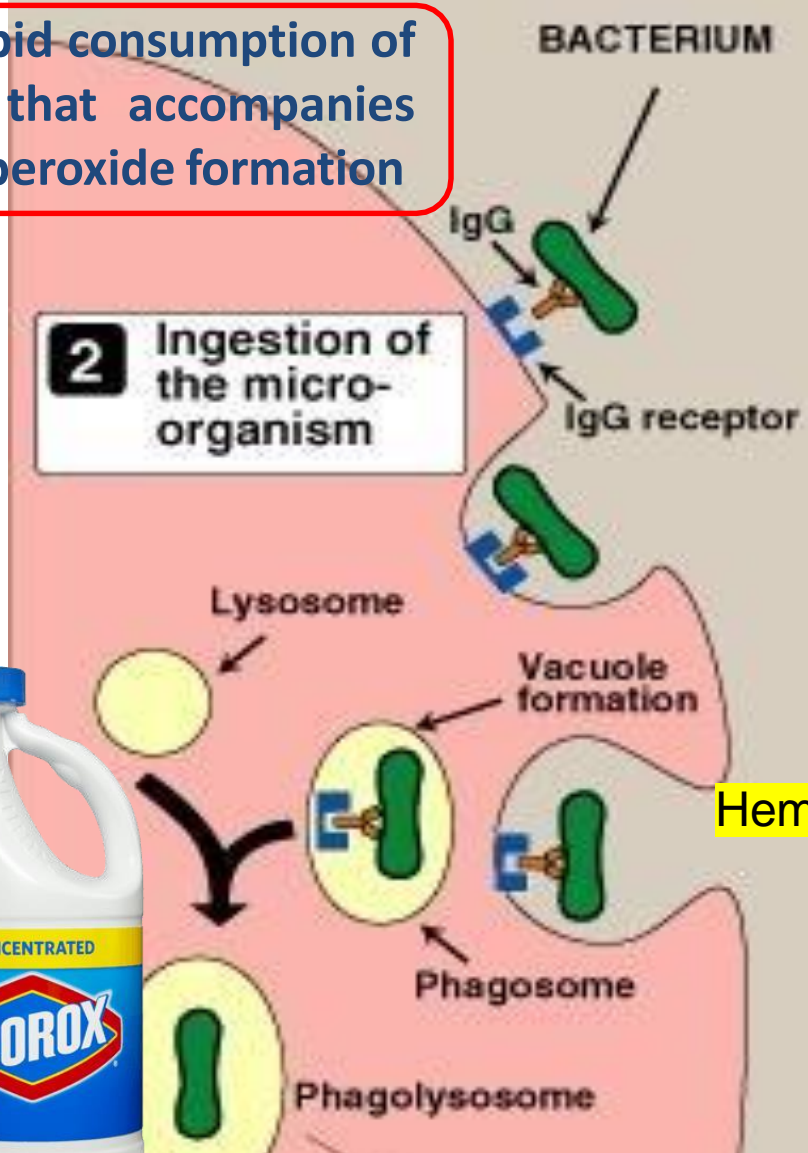
- **The complement in this slide: :** Co-enzyme Q receives electrons from complex 1 and 2, these electrons can be added to oxygen because they are in oxygen rich environment to form superoxide O_2^- as a byproduct.

Phagocytosis; the oxygen dependent pathway of microbial killing by WBCs

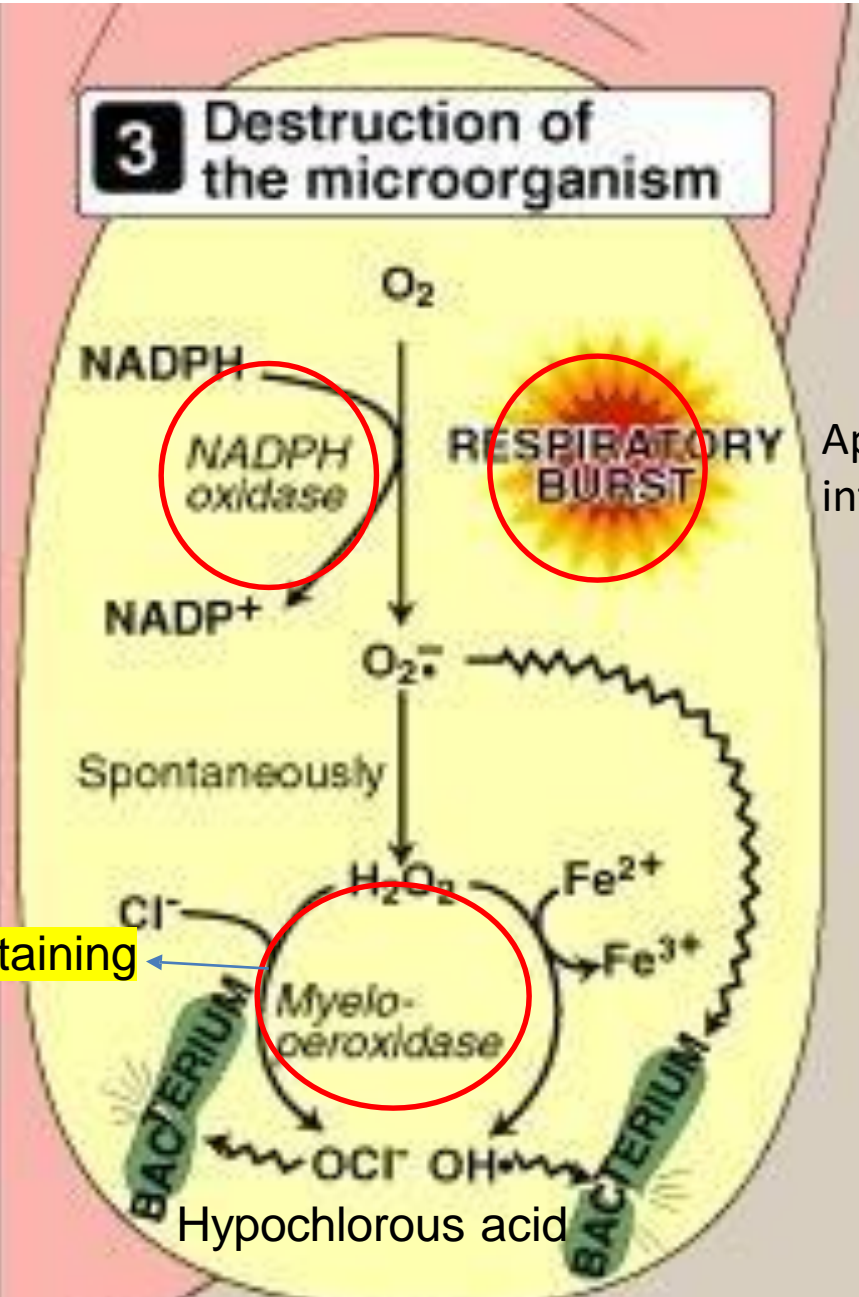
1 Attachment of the pathogen to a phagocytic cell

Rapid consumption of O_2 that accompanies superoxide formation

2 Ingestion of the micro-organism



3 Destruction of the microorganism

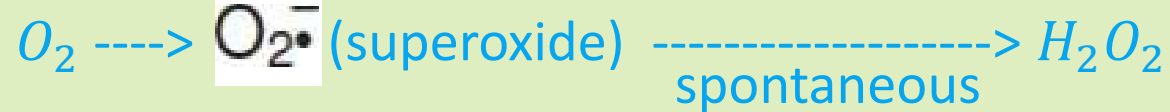


Appears during infections

Heme containing

H_2O_2 can also be reduced to water by catalase or glutathione peroxidase

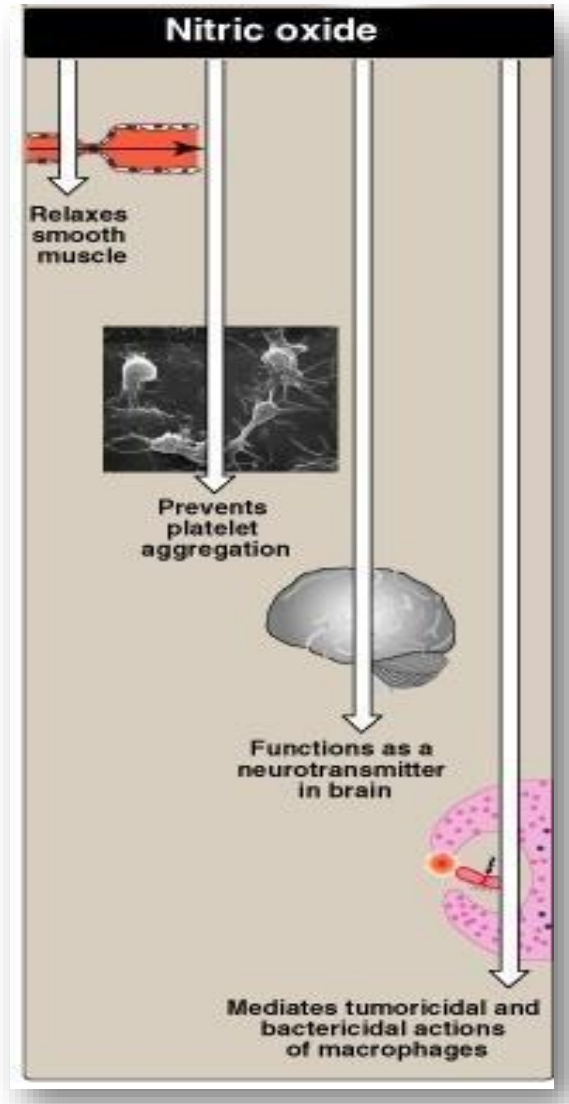
■ **The complement in this slide:** An invading bacterium is recognized by the immune system (IgG) and attacked by antibodies that bind it to a receptor on a phagocytic cell. Inducing the ingestion of the bacterium in cell forming **phagosome**. Then it will fuse with lysosome forming **phagolysosome**, This will induce the **respiratory burst** (kind of reaction makes ROS and NROS), then NADPH oxidase is activated which leads to:



Then H_2O_2 can bind with Cl^- in the presence of MPO (Myeloperoxidase) $\rightarrow OCl^-$ Hypochlorous acid

OR H_2O_2 can accept an electron from Fe^{+2} , to form OH^-

NOTE: all of them ROS and NROS (H_2O_2 , OCl^- , OH^- , $O_2^{\bullet -}$) can destruct the bacteria.
 ALSO, H_2O_2 can also be reduced to water by catalase or glutathione peroxidase.



NO and Reactive Nitrogen Oxygen Species (RNOS)

- Diffuses readily
- Essential for life and toxic
- Neurotransmitter , vasodilator (by relaxation of smooth muscles)
- ↓Platelet aggregation
- At high concentration combines with $O_2^{\bullet-}$ or O_2 to form **RNOS**
- **RNOS** (if there concentration increase) are involved in neurodegenerative diseases and inflammatory diseases

Read the picture important!!!!

- **The complement in this slide:** NO is a signaling molecule that is produced in the cell and is also considered as a ROS. It's a small gaseous molecule so it's **diffused very easily**. Therefore, it's effect is wide, and it may cause **toxicity in higher amounts**, because it is source of RNOS.
- How is NO produced ?
- It's produced from arginine amino acid (because it contains a plenty of Nitrogen atoms) and that's through NO synthase with the interference of NADPH as indicated in the slide.
- NO synthase has different isoforms :
- 1- nNOS "neural" . 2- eNOS "endothelial". (they are **always active and expressed**)
- 3- iNOS (inducible -**need stimulus**- Ca^{+2} independent) which is activated under certain conditions, one of these conditions is the process of fighting an infection (to kill a microorganism) which will induce the production of RNOS to **kill invading bacteria**.

NO Synthesis

■ **NOTE:** This reaction consumes NADPH, so it is bad when it happens alot

NO Synthase

Three isoforms

nNOS neural

eNOS endothelial

Both are constitutive

iNOS inducible Ca^{+2}

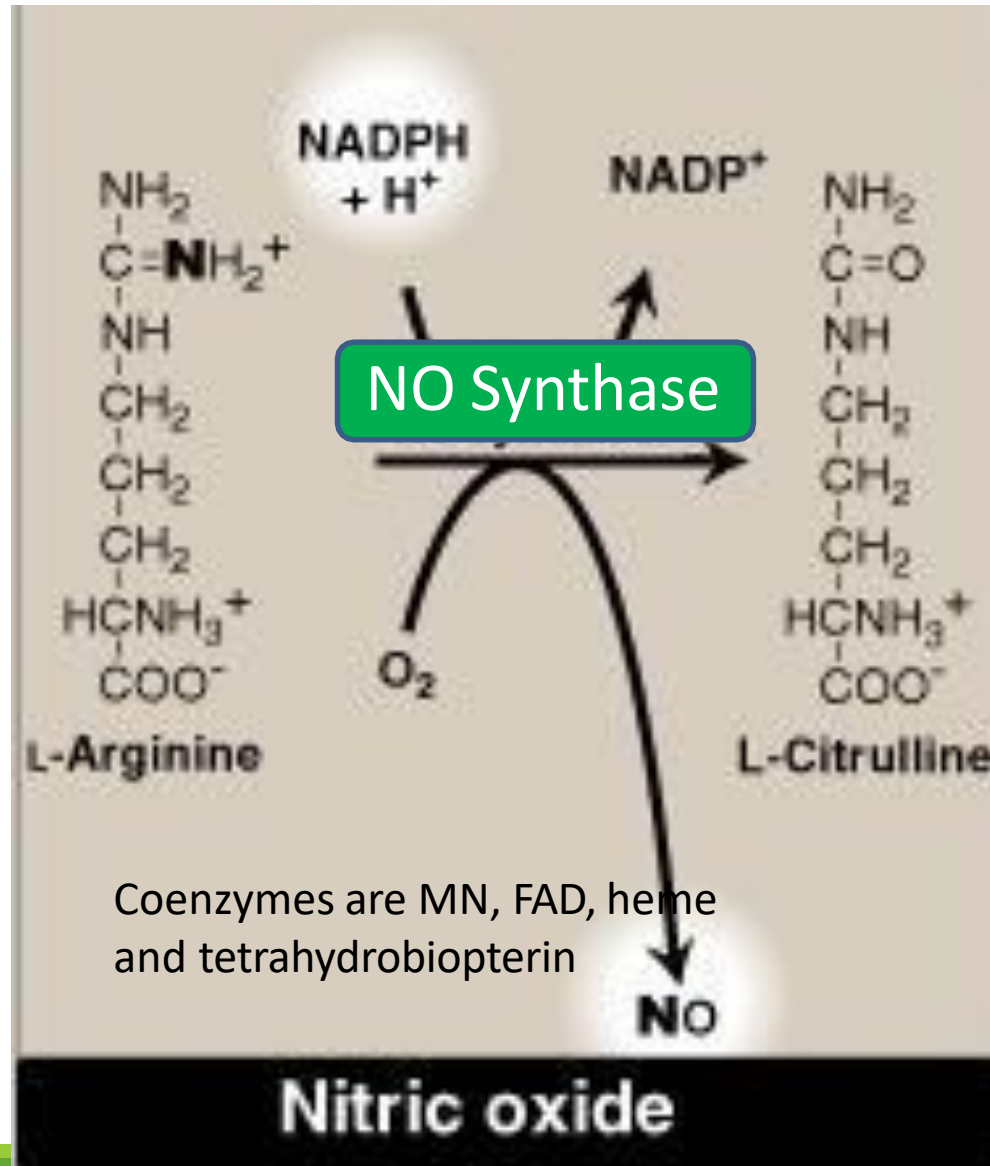
independent

Induction of transcription in

many cells of immune

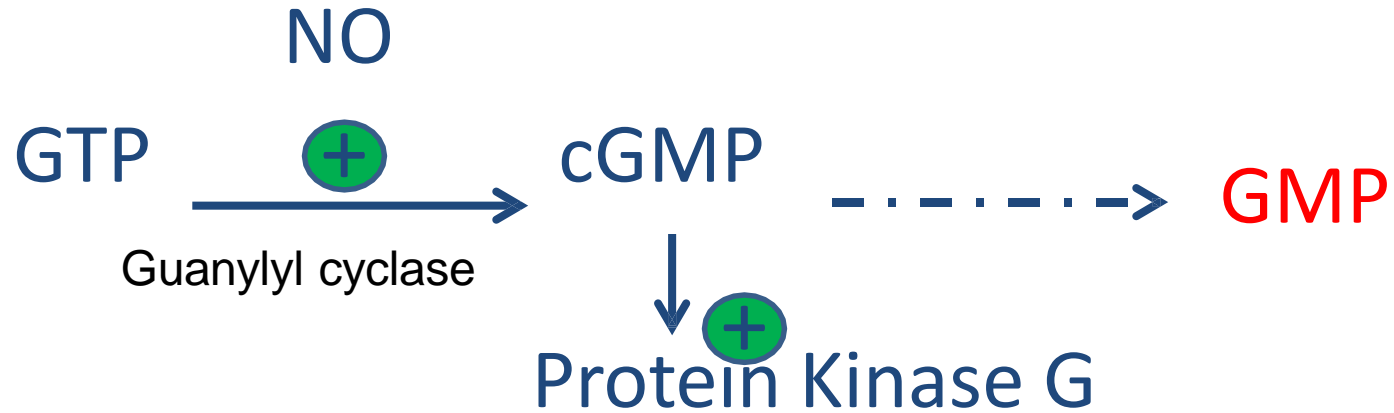
system $\rightarrow \uparrow \uparrow NO \rightarrow$ RNOS

to kill invading bacteria



Action of NO on vascular endothelium

Synthesis by endothelia cells \rightarrow smooth muscle



Phosphorylation of Ca^{2+} channels

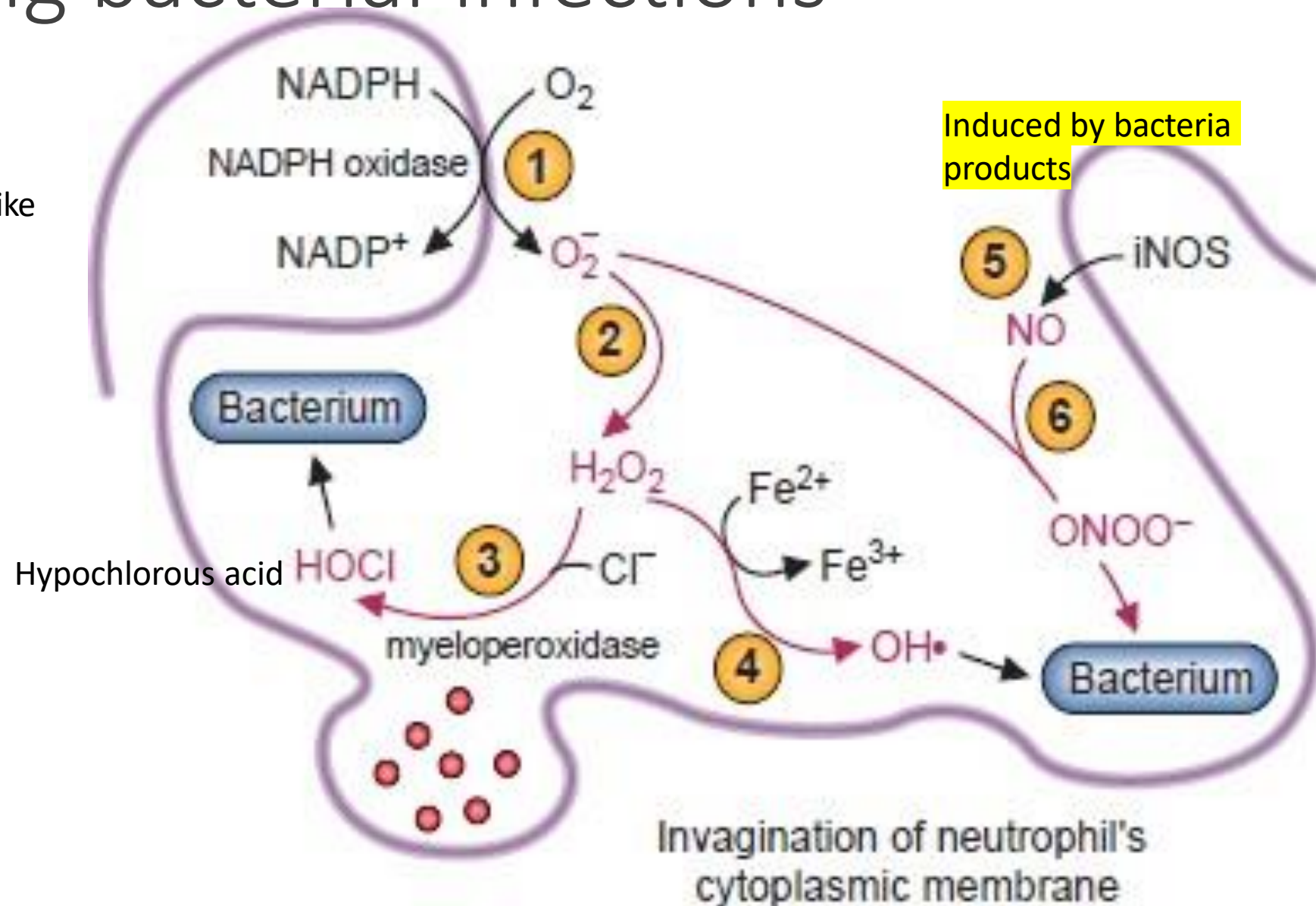
NOTE: channels will close

$\downarrow\downarrow \text{Ca}^{2+}$ entry into smooth muscle cells and causes muscle relaxation and lowers blood pressure

NO role during bacterial infections

Activated cells like macrophages

- **NOTE:** from 1 – 4 we took previously, last 2 points 5, 6 briefly:
- **iNOS** is induced by bacteria → **NO** synthesis → **ONOO⁻** → destruct bacterium



Invagination of neutrophil's cytoplasmic membrane

Insulin Resistance (IR)

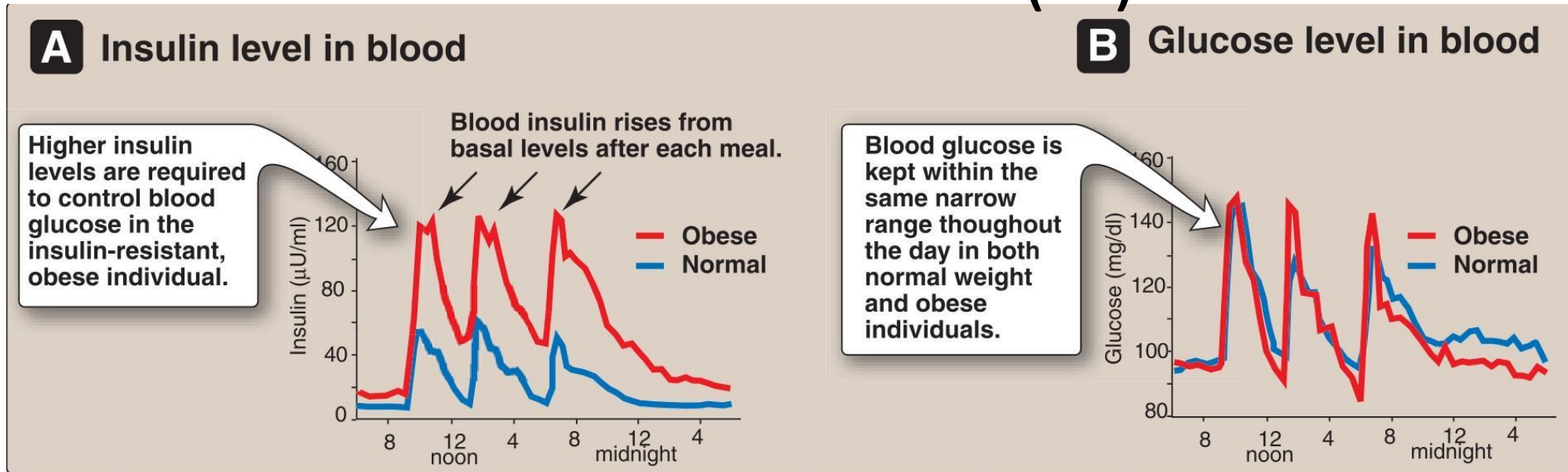


Figure 25.7

Blood insulin and glucose levels in normal weight and obese subjects.

- ✓ IR is the decreased ability of target tissues, such as liver, adipose, and muscle, to respond properly to normal (or elevated) circulating concentrations of insulin.
- ✓ IR is characterized by uncontrolled hepatic glucose production, and decreased glucose uptake by muscle and adipose tissue.
- ✓ Obesity is the most common cause of IR
- ✓ IR alone will not lead to type 2 diabetes.
- ✓ Type 2 diabetes develops in insulin-resistant individuals who also show impaired B-cell function

- **The complement in this slide: IR happens in patients with diabetes type 2 (in adults)**
- **Type 1:** in children, there might be dysfunction of Pancreatic cells because of autoimmune disease, so the pancreas will not produce insulin or makes very little amount (you can take insulin as a therapeutic, but it is more serious than type 2).
- **Type 2** usually happen to **obese people**.
- When u eat a large meal (even if you are not obese) the pancreas will release **large amount of insulin** Proportional to glucose concentration.
- The more sugar in the blood, there will be increase in insulin secretion , this will lead to draining of the pancreas(asking him to work a lot), in the end whatever the person is normal or obese he will reach fasting blood sugar after 2 hours.
- The same amount of insulin **doesn't produce** the same effect, like **Tolerance**
- And this will exhaust the pancreas.
- **NOTE:** When insulin levels are elevated (obese people) --> bodily cells develop resistance to insulin.

لما يضل الجسم يفرز كميات كبيرة من الأنسولين وبصورة متكررة و(خصوصا الشخص السمين) فيؤدي الجسم الى افراز كميات أكبر من الأنسولين لإدخال نفس كمية السكر، وهذا يسمى ب مقاومة الأنسولين

V2 slide 19

OR H_2O_2 can accept an electron from Fe^{+2} , to form OH^-

ALSO, H_2O_2 can also be reduced to water by catalase or glutathione peroxidase.